

LETTERS TO THE EDITOR

Acylation of Per-6-*O*-(*tert*-butyl)(dimethyl)silyl- α -cyclodextrin with Acetylsalicylic Chloride

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Earlier, aiming to synthesize the conjugates of α - and β -cyclodextrins containing drug residues at the cyclodextrin torus side occupied with secondary hydroxyl groups, we have performed acylation of per-6-*O*-(*tert*-butyl)(dimethyl)silyl derivatives of α - (**I**) and β -cyclodextrin (**II**) at the second hydroxyl groups with chloroanhydrides of a series of aromatic monocarboxylic acids, followed removal of protecting silyl groups [1, 2]. Acylation of the silyl derivatives **I** and **II** occurred with good yields in pyridine as well as (in the case of compound **II**) in DMF in the presence of *N,N*-dimethylaniline (method *a*) or C_6H_6 (method *b*) in the presence of Et_3N , the amines acting as the formed hydrogen chloride scavengers (Scheme 1).

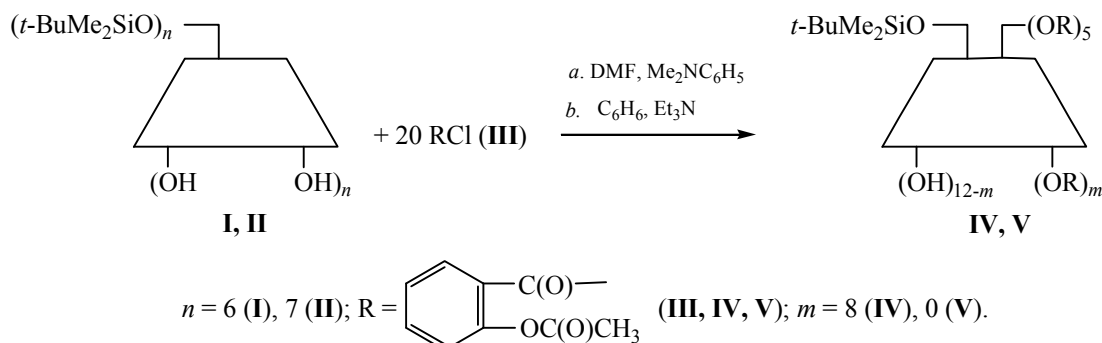
The trials to perform analogous acylation of the silyl derivative of α -cyclodextrin **I** with acetylsalicylic chloride **III** using the methods *a* and *b* unexpectedly led to the products of substitution of a substantial part of the protecting silyl groups with acetylsalicylic acid

residues. From the 1H NMR spectroscopy data, under the acylation conditions only one silyl group was preserved and eight (**IV**, $m = 3$, method *a*) or five (**V**, $m = 0$, method *b*) of the acetylsalicylic acid residues were attached to the cyclodextrin backbone.

Unexpected desilylation of the α -cyclodextrin derivative under the conditions of acylation with acyl chloride **III** using methods *a* and *b*, different from conventional acylation in pyridine [2], was apparently due to supramolecular effect of the inner cavity of α -cyclodextrin, as we have mentioned earlier discussing the silyl derivatives of cyclodextrins [3, 4].

Methods of acylation. *a.* A solution of 0.79 g of acyl chloride **III** in 4 mL DMF was added dropwise upon stirring at $0^\circ C$ during 30 min to the solution of 0.33 g of the silyl derivative **I** and 0.53 g of *N,N*-dimethylaniline in 3 mL DMF. The mixture was incubated during 24 h at $20^\circ C$, then evaporated in vacuum (10 mmHg) to oily state; the residue was

Scheme 1.



washed with 10 mL diethyl ether, dried in vacuum, and thoroughly triturated with 10 mL of H₂O. The solid product was separated off and dried at 1 mm Hg and 80°C during 3 h. Yield of compound **IV** 0.324 g (68%), mp 144–146°C (decomp.), *R*_f 0.8. ¹H NMR (CDCl₃), δ , ppm: 0.03 s [6H, Si(CH₃)₂], 0.89 s [9H, C(CH₃)₃], 1.62–2.11 m [24H, C(O)CH₃], 2.80–4.51 m (36H, C²H–C⁶H), 4.96 m (6H, C¹H), 5.05–5.25 br.s (9H, C²OH), 7.04–8.09 m (32H, C₆H₄). Found, %: C 57.53; H 5.10. C₁₁₄H₁₂₂O₅₄Si. Calculated, %: C 57.43; H 5.16.

b. The synthesis was performed as described in procedure *a*, from 0.33 g of the silyl derivative of **I**, 0.445 g Et₃N in 3 mL C₆H₆ and 0.79 g of acyl chloride **III** in 4 mL C₆H₆. The formed precipitate of triethylamine hydrochloride was filtered; the filtrate was evaporated to oil in vacuum, thoroughly triturated with 10 mL of diethyl ether, filtered off, and in vacuum. Yield of compound **V** 0.231 g (61%), mp 162–165°C (decomp.), *R*_f 0.7. ¹H NMR (CDCl₃), δ , ppm: 0.03 s [6H, Si(CH₃)₂], 0.90 s [9H, C(CH₃)₃], 2.11–2.30 m [15H, C(O)CH₃], 2.82–4.53 m (36H, C²H–C⁶H), 4.70 m (6H, C¹H), 5.01–5.21 br. s (12H, C²OH), 7.04–7.93 m (32H, C₆H₄). Found, %: C 55.10; H 5.45. C₈₇H₁₀₄O₄₅Si. Calculated, %: C 55.06; H 5.52.

¹H NMR spectra were recorded using the Jeol-ECX400 spectrometer (400 MHz). TLC analysis was

carried out on alumina plates Silufol UV-254 with fixed silica layer; eluent: 6% NH₃ in H₂O–BuOH–EtOH 5 : 4 : 5.

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